



**NATIONAL OPEN UNIVERSITY OF NIGERIA**  
**UNIVERSITY VILLAGE, PLOT 91 CADASTRAL ZONE, NNAMDI AZIKIWE**  
**EXPRESS WAY,**  
**JABI - ABUJA.**  
**FACULTY OF SCIENCES**  
**DEPARTMENT OF CHEMISTRY**  
**2025\_2 EXAMINATIONS**

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**COURSE CODE:** CHM 408  
**COURSE TITLE:** POLYMER CHEMISTRY  
**COURSE UNIT:** 2  
**TIME:** 2 HOURS  
**INSTRUCTION:** *Answer question one and any other two questions.*  
*Question 1 score 30marks with the other two scoring 20marks each*

**Question 1**

- a. Discuss Addition polymerization? (6mark)
- b. Discuss the following terms used in describing the response of material to stress –strain
  - i. Ultimate elongation(3marks)
  - ii. Tensile strength(3marks)
- c. Discuss Hydrolytic degradation? (6mark)
- d. Explain structural isomerism? (6mark)
- e. Discuss the following;  
Nuclear Magnetic Resonance Spectroscopy (NMR)(3mark)  
X-ray Diffraction (XRD)(3mark)

**Question 2**

- a. Discuss Geometrical isomerism(4mark)
- b. With the aid of a table, Show the condensed formula and a physical property for the below polymers(6mark)
  - i. Cis-Polychloroprene
  - ii. Poly(methyl methacrylate)
  - iii. Poly(vinylacetate)
  - iv. Polyacrylonitrile
- c. Give four polymers that are gotten from anionic polymerization(2mark).
- d. For solution process, show the defined expressions for free energy, enthalpy and entropy of mixing(3mark)
- e. Explain how the below affect the mechanical strength of a polymer? (5marks)
  - i. Effects of Additives
  - ii. Effect of Blending and Copolymerization

### Question 3

- a. From the first and second law of thermodynamics derive the pressure-volume expansion and work relationship. (6mark)
- b. Briefly discuss cationic polymerization stating the stages of reaction (6mark)
- c. For solution process, show the defined expressions for free energy, enthalpy and entropy of mixing (3mark)
- d. Discuss the below types of copolymerization
  - i. Random copolymers (2.5mark)
  - ii. Alternating copolymers (2.5mark)

### Question 4

- a. What is dynamic mechanical thermal analysis (4marks)
- b. Discuss how the below properties affect solubility of polymers.
  - i. Crystallinity (strong hydrogen bonding) (2.5mark)
  - ii. Polarity or non-polarity (2.5mark)
- c. Discuss five benefits of polyurethane (8mark)
- d. Discuss the use of IR spectroscopy to identify polymers; (3mark)

### Question 5

- a. Give a brief information about polymer phase equilibrium (7mark)
- b. In a tabular form, show three applications of any two thermosetting polymers (5mark)
- c. Discuss the following complex instrumental techniques used to identify polymers
  - Glass Liquid Chromatography (GLC) (4mark)
  - Gel Permeation Chromatography (GPC): (4mark)