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NATIONAL OPEN UNIVERSITY OF NIGERIA PLOT 91, CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESSWAY, JABI – ABUJA FACULTY OF SCIENCES DEPARTMENT OF CHEMISTRY 2023_1 POP EXAMINATION.

COURSE CODE: CHM 409 COURSE TITLE: PHOTOCHEMISTRY & PERICYCLIC REACTIONS COURSE UNIT: 2 INSTRUCTION: Answer question 1 and any other two questions Time : 2 hours

QUESTION 1

- 1) (a) Briefly explain the following terms / concepts
 - (i) Antarafacial and suprafacial cycloaddition reactions (6 marks)
 - (ii) symmetry allowed and symmetry forbidden pericyclic reactions (6 marks)
 - (iii) Photosensitization (3 marks)
 - (iv) Optical pumping in Laser Technology. (3 marks)
- (b) Write a general equation that represent photodissociation of X_2 molecule. (2 marks)
- (c) Use Einstein law to show that the energy of radiation can be written as
- $E = \frac{1.1925 \times 10^5}{\lambda} kJ/mole$. Hence calculate the energy that can be generated by a radiation whose wavelength is 10 pm (10 morks)

whose wavelength is 10 nm (10 marks)

QUESTION 2

- 2) (a) State the Woodward-Hoffmann's rule for pericyclic reactions (5 marks)
 - (b) Predict the stereochemistry of 5, 6 dimethylcyclohexa-1,3-diene when formed (i) thermally, and (ii) photochemically, from trans, cis trans-octa-2,4,6-triene. (15 marks)

QUESTION 3

- 3. Briefly explain using an example in each case the following terms
 - (a) Ring-closure electrocylic reaction (6 marks)
 - (b) Dyotropic rearrangements (6 marks)
 - (c) sigmatropic rearrangement reactions (8 marks)

QUESTION 4

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4. (a) Highlight the mechanism involve in the photochemical reaction of toluene with

Chlorine (6 marks)

- (b) When irradiated with light of 5000 Å wavelength, 1×10^{-4} mole of a substance is decomposed. How many photons are absorbed during the reaction if its quantum efficiency is 100. (Avogadro number = 6.02×10^{23}). (11 marks)
- (c) What are cyclo addition reactions (3 marks)

QUESTION 5

5. (a) List the three (3) major components of laser (6 marks)

(b) (i) Photolysis of HBr is known to be characterised with two sets of reaction. Name the reaction and write suitable equations for each. Hence write the overall reaction. (8 marks)

(ii) Write an equation that can be used to calculate the quantum yield of a photochemical reaction. Hence calculate the quantum yield for photolysis of HBr. (6 marks)