

150g of NaOH in 1000cm<sup>3</sup> of water is said to be a?

Concentrated solution

Which of the following is considered while working in the laboratory?

Precaution

The number of moles of solute in one kilogram (kg) of solution is called ?

Molality

What is prepared by weighting a pure solute, dissolve it in a suitable solvent, usually water and make up the solution to a definite volume in a volumetric flask.?

Standard solution

The number of moles of solute in one kilogram (kg) of solution is known as?

Molality

Find the mole of NaOH if its mass is 40g, (Na = 23, O = 16, H =1).

1.00 mol

What volume of it must be diluted with water to obtain 500cm<sup>3</sup> of 0.01 molar solution. If the stock solution is 1.0 M?

5cm<sup>3</sup>

Which of the following is not a unit of concentration?

Litre/mol

Which of the following is the most common accidents in chemistry laboratory?

cut

Which of the following is not correct?

eating or drinking is allowed in the laboratory

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