

1. P - block elements are elements in groups _____.
13 to 18
2. S- block elements are _____.
group 1 and 2 elements
3. If the bond length in a XY molecule is 212, what will be the covalent radius of atom X, if the covalent radius of Y atom is 93.
2.27
4. The first element on the periodic table is _____.
hydrogen
5. Normal representative or main group elements are _____.
the s and p block elements
6. Electronegativity _____ of S and P block elements.
increases across the period
7. The s- block elements are elements in which the _____ enters the s atomic orbital last electron
8. D and F - block elements are also known as _____.
transition and inner transition elements
9. The energy required to remove the least strongly bond electron from an isolated gaseous atom in its ground state is known as the _____.
ionization energy
10. Which of this statement is correct _____?
Atomic radius decreases across a period because, across the period effective nuclear charge increases, exerting greater force on the outer most electron, hence pulling the electron cloud of the outer most shell inward nearer the nucleus.

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