

CHM201

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1. From theoretical perspective, a generalization which seems to explain the _____ existing between the phases in a system is called phase rule

reaction

reactivity

--->> equilibrium

density

2. Reactions between two or more reactants depends on _____

--->> effective collision

scalar collision

vector collision

circular collision

3. The process where solid changes to gaseous state is called _____

--->> sublimation

freezing

condensing

boiling

4. larger surfaces give room for _____ in liquids

boiling

--->> evaporation

critical point

triple point

5. The pressure exerted by gas molecules in a container is as a result of _____ on the wall of the container

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escaping

entering

diffusing

6. Smaller molecules diffuses _____ than larger molecules

slower

easier

--->> faster

moderately

7. The change of state involve _____ changes

atomic

--->> heat

pressure

gas

8. When we have a second order phase transition, the first derivative of _____ with respect to temperature is continuous

--->> chemical potential

decline function

circular function

discrete function

9. The sum of enthalpy of vaporization and the enthalpy of fusion gives the enthalpy of _____

condensation

--->> sublimation

boiling

freezing

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potential energy

circular energy

straight energy

--->> kinetic energy

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