

CHM101

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1. The reduction of aldehydes, ketones, organic acids and their derivatives to give alcohols is made possible by the use of _____

--->> Lithium aluminium hydride

lithium sulphide

lithium ore

lithium oxide

2. The high boiling and melting points of water is attributed to _____

--->> hydrogen bonding

covalent bonding

dative bonding

ionic bonding

3. Alkali metals are very good ____ of heat and electricity

source

distribution

--->> conductor

neutralizer

4. A very useful fluid to life that has iron(Fe) as a central metal atom is called _____

mucus

body water

--->> haemoglobin

synovial fluid

5. Boron hydride is an example of _____ covalent hydrides

trimeric

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single

monomeric

6. Consider the elements : Br, Ca, Na, Mg. Which of the these elements is most electronegative ?

--->> Br

Mg

Ca

Na

7. A group of ions or molecules surrounding a central metal atom are called _____

--->> Ligands

adsorbent

addendum

adduct

8. metallic hydrides are _____ than the parent metal.

highly viscous

more dense

more coloured

--->> less dense

9. Bright yellow light on the street are made up of _____

Aufbau light

manganese light

--->> sodium light

oxygen light

10. Sodium trioxocarbonate is used in _____ powder

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--->> baking

soluble

adsorbent

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