

sulphuric acid

[SED226] The quality of sulphuric acid made or consumed by a country has been used as a measure of the state of its economy

[SED226] The key components of heavy industry is industrial processes

[SED226] Heating sulphide ores like pyrite in an excess of air produces sulphur dioxide

[SED226] The process in which nitrogen reacts with hydrogen to produce ammonia is referred to as Haber process

[SED226] The industrial process used in the manufacture of sulphuric acid is Contact process

[SED226] Sulphur dioxide can be made by burning sulphur in an excess of air

[SED226] The major use of nitrogen in the Haber process is the production of ammonia

[SED226] The compromise temperature that produces a reasonably high proportion of ammonia in a very short time is 400-450°C

[SED226] The procedures involved in industrial processes include the following except cultural procedures

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