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radiationncovers the wavelength range from about 100 nm - 800 nm

[CHM309] The atomic absorbtion spectrophotometer has major components three
[CHM309] The energy of the most stable electron configuration ofnan element is its ground state is lowest energy
[CHM309] The interaction of light with matter is spectroscopy
[CHM309] In $c = v\tilde{A}\check{Z}\hat{A}$ », the term $\tilde{A}\check{Z}\hat{A}$ » represents the distance between two successive peaks
[CHM309] The molecular orbitals with electron density above and below the plane of the bonded nuclei are called pi orbitals
[CHM309] Photons are characterised by their energy
[CHM309] Molecular orbital with electron density in the plane of the bondednnuclei, that is, along the axis connecting the bonded nuclei, are calledn sigma orbital
[CHM309] The ground state configuration of an atom is predicted from the Atomic orbital principle
[CHM309] Standard unit of wavelength is nanometer

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