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Molecular Spectroscopy are associated with changes in the following EXCEPT changes in the configuration of electron Ammonia (NH3) shows\_\_\_\_\_ hybridisation sp3 Boron triflouride,BF3, shows \_\_\_ hybridisation sp2 diative transition can be classified and characterised by multipolarity, which of these transition will give the highest Einstein coefficient, A (S-1) Electric dipole Which of these energies is defined by Internal energy of a molecule All of the above When any of the components of electromagnetic radiation (EMR) interacts with matter, the energy level of the molecule may change through interaction with Proton If the dipole moment of hydrogen molecule is zero then its transition moment when it interacts with electromagnetic radiation Zero Which of these temperatures show that a particular type of energy is not a significant component of molecular spectroscopy 2500K Which of these is NOT a limitation of valence bond Theory The Valence bond theory do not concentrate on the valence electron Types of interaction that can be encountered when atoms are close to each other **INCLUDE all EXCEPT** Repulsion between electrons

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