

titration reduction?  
chemical equilibrium is achieved

[CHM202] The end point of redox titration is determined?  
calorimetrically

[CHM202] Which of the following option is correct about normality  
meq/L

[CHM202] In volumetric calculation the key step is to relate the moles of titrant to the  
mole of?  
analyte

[CHM202] Which of the following is not a type of volumetric procedures?  
substitution titration

[CHM202] What substance takes an electron in redox reaction?  
oxidizing agent

[CHM202] Millimole is equal?  
 $\text{mL} \times \text{M}$

[CHM202] Bronsted-Lowry described a base as a substance that can?  
accept proton

[CHM202] What type of titration involves adding a known excess of the standard  
reagent to the analyte?  
back titration

[CHM202] Mole can be define as?  
mass divided by molar mass

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