

concentration, represent ____ order reaction
first

[CHM201] In natural gas production industry, the real gas equation most commonly used is ____
 $PV = ZnRT$

[CHM201] The particular temperature at which the surface between the two phases (liquid and vapour) disappears is referred to as ____
critical temperature

[CHM201] Consider the equation, $dU = C_v dT$ is the heat capacity at ____
constant

[CHM201] A typical electrode compartment is made up of an electrode and its ____
electrolytes

[CHM201] The expression that show how the initial concentration decreases with time is ____
rate law

[CHM201] A condition when three different phases(solid, liquid, and gas) all simultaneously co-exist in equilibrium is called ____
triple point

[CHM201] The expression, $H = U + PV$ is the definition of ____
enthalpy

[CHM201] From the assumptions of the kinetic theory of gases, the absolute ____ of a gas is a measure of the average kinetic energy of all the molecules.
temperature

[CHM201] A notable assumption of the kinetic theory is that all gases are made up of tiny particles called ____
molecules

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