

atom in its ground state is called ____
electron affinity

[CHM101] The trend in the periodic table is that atomic radius of s- and p- block elements ____ down the group.
increases

[CHM101] Electrons associated with solvent in the solutions of alkali metals in liquid ammonia is termed ____
solvated electrons

[CHM101] The energy required to remove an electron completely from an atom of an element is termed ____
first ionization energy

[CHM101] The electronic configuration of Fe atom is ____
\$\$[Ar] 4s^{\{2\}} 3d^{\{8\}}\$\$

[CHM101] The greater the effective nuclear charge, the ____ the electron affinity of the atom.
higher

[CHM101] The unit of electron affinity is ____
KJ/mol

[CHM101] Across the period, the size of an atom decreases and effective nuclear charge ____
increases

[CHM101] The higher the effective nuclear charge, the greater the ____ exerted by the nucleus on the added electron
force of attraction

[CHM101] The tendency of an atom to attract electrons to itself when combined in a compound is called ____
electronegativity

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