

dual

[PHY202] The isotopes of an element all have the same atomic number

[PHY202] The weakest of the four fundamental interactions is the gravitational

[PHY202] It is impossible for two electrons with the same spin quantum number to be in the same orbital. According to Pauli exclusion principle

[PHY202] According to Pauli exclusion principle. Two electrons in the same orbital have opposite spins

[PHY202] What parts of its structure are chiefly responsible for an atom's mass? nucleus

[PHY202] The atomic number of an element is the number of protons in its nucleus

[PHY202] The property of an electron which gives rise to its angular momentum about an axis within the electron is its electron spin

[PHY202] Nearly all the volume occupied by matter consists of empty space

[PHY202] Maximum number of electrons that can be accommodated in a shell is given by  $2n^2$

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