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NATIONAL OPEN UNIVERSITY OF NIGERIA UNIVERSITY VILLAGE, PLOT 91 CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESS WAY, JABI - ABUJA.

FACULTY OF SCIENCES

DEPARTMENT OF PURE AND APPLIED SCIENCES

FEBRUARY/MARCH2018 EXAMINATION

COURSE CODE: CHM 302

COURSE TITLE: POLYMER CHEMISTRY 1

TIME: 2 HOURS

INSTRUCTION: Question one is compulsory. Answer question one and

any other three questions.

QUESTIONONE

- 1ai)Polymers are macromolecule while all macromolecules are not polymers. Explain. 3 marks
- 1aii) Enumerate on the classes of polymers.11 marks
- 1b) Expatiate on condensation polymerization.5marks
- 1c) Account for the principle of liquid crystal nature of polymers.3 marks
- 1d) Explain briefly crystallinity in polymers.3 marks

QUESTION TWO

2a) Write short note on addition polymerization including the mechanism of its formation.

15 marks

QUESTION THREE

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in a polymer determine the properties, shapes and uses of the following:

- i. Thermosets
- ii. Thermoplastics

12 marks

3b) Will cross-links be possible in polyalkanes? Give reason for your answer.

3 marks

QUESTION FOUR

Enumerate on the effect of chain length on the physical properties of polymers.

9 marks

Highlight the main differences between elastomers and fibers.6 marks

QUESTION FIVE

- 5a) Write short notes on the following types of linkages in polymers;
 - i. Linear alternating copolymers
 - ii Random or statistical copolymers
 - iii Block copolymers

11 marks

- 5b) Complete the following polymerization reactions;
- 1. nCH₂= CH₂O₂ initiator 2 marks 1500atm, 2000⁰C
 - 2. HOOC[CH₂]₄COOH + H₂N[CH₂]₆NH₂-2H₂O2 marks