



**NATIONAL OPEN UNIVERSITY OF NIGERIA**  
**PLOT 91, CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESSWAY, JABI - ABUJA**  
**FACULTY OF SCIENCES**

**DEPARTMENT OF PURE AND APPLIED SCIENCE**

**2021\_2 EXAMINATIONS.**

**COURSE CODE:** PHY311  
**COURSE TITLE:** KINETIC THEORY AND STATISTICAL MECHANICS  
**CREDIT UNIT:** 2  
**TIME ALLOWED:** (2 HRS)

**INSTRUCTION:** *Answer question 1 and any other three questions*

**QUESTION 1**

(a) Define the following terms; (i) Most probable velocity (ii) Root-mean-square velocity  
[5 marks]

(b) Show that the most probable speed at which  $n(v)$  has its maximum value  $V_p = \sqrt{\frac{2RT}{M}}$   
[10marks]

(c) Show that the root mean square speed of gas molecule  $V_{rms} = \sqrt{\frac{3RT}{M}}$  [10marks]

**QUESTION 2**

(a) Derive Dulong-petit's law on the basis of equipartition theorem [5marks]

(b) Show that for a perfect gas represented by a grand canonical ensemble, the probability of finding the subsystem with  $n$  atoms is given by Poisson distribution

$\omega(n) = \frac{1}{n!} (n)^n e^{-\bar{n}}$  where  $\bar{n}$  is the number of atoms present [10marks]

**QUESTION 3**

(a) If twelve particles are distributed randomly between two boxes A and B with equal probability, then calculate

(i) The probability of the distribution (8, 4)

(ii) The probability of the most probable distribution

(iii) The probability of least probable distribution. [9marks]

(b) Find the probability that in tossing a coin 12 times we get (i) 4 heads 8 tails (ii) 6 heads 6 tails  
[6marks]

**QUESTION 4**

Three particles are to be distributed in four energy states  $a, b, c$  and  $d$  write down all the possible ways for such a distribution if the particles are (i) Fermions (ii) Bosons [15marks]

**QUESTION 5**

Let  $v_x, v_y, v_z$  represent the three Cartesian components of velocity of a molecule in a gas. Using symmetry consideration and equipartition theorem, deduce, expression for the following mean values in terms of K, T and m. (i)  $\langle v_x \rangle$  (ii)  $\langle \bar{v}_x^2 \rangle$  (iii)  $\langle v_x v_z \rangle$  (iv)  $\langle (v_x + bv_y)^2 \rangle$

[15marks]