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NATIONAL OPEN UNIVERSITY OF NIGERIA DEPARTMENT OF PURE AND APPLIED SCIENCES 2020_2 EXAMINATIONS...

COURSE CODE: CHM 309
COURSE TITLE: Applied Spectroscopy
TIME: 3 HRS

INSTRUCTION: Answer question 1 and any other 4 questions

QUESTION 1

- (a)Discuss the following factors that affect the absorption of radiation in the UV/visible region
- (i) Solvent Effects (4 marks)
- (ii)Substituent Effects (4 marks)
- (b) Explain the concept of molecular ion in mass spectroscopy (4 marks)
- (c) Explain three (3) factors that determin the intensity and energy level of absorption in IR spectroscopy (3 marks)
- (d) Discuss the concept of chemical ionization in mass spectrometry (4 marks)
- (e) Explain the term chemical shift in Nuclear magnetic Resonance (NMR) spectroscopy (3 marks)

Question 2

- (a) Discuss the theoretical basis of Nuclear Magnetic Resonance (7 marks)
- (b) Explain the spinning Nuclear of hydrogen atom (3 marks)
- (c) State the universally accepted reference used in the chemical shift measurement. (2 marks).

Question 3

- (a) Discuss the kinds of transitions exhibited by electrons in a molecule in the UV visible spectra. (7 marks)
- (b) Explain the anti-bonding orbitals in a molecule in the UV visible spectra. (5 marks)

Question 4

- (a) Show the relationship between absorbance and concentration of the sample as stated in Beer Lambert law. (9 marks)
- (b) Explain the relationship between energy of photon and the wavelength. (3 marks)

Question 5

(a) Explain positive ion chemical ionization (PICI) as one of the ionization techniques used in GC-MS. (4.5Marks).

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(c) state the relationship between the magnetic moment, energy, and frequency. (2.5 marks)

Question 6

- (a) State the five main components of a typical NMR spectrometer (7.5 marks)
- Explain the principles behind thermospray as one of the interfaces used in LC-MS (4.5 marks) (b)