



NATIONAL OPEN UNIVERSITY OF NIGERIA
PLOT 91, CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESSWAY, JABI - ABUJA
FACULTY OF SCIENCES
DEPARTMENT OF PURE & APPLIED SCIENCES
SEPTEMBER, 2020_1 EXAMINATION

CHM 305 ORGANIC CHEMISTRY III

CREDIT UNIT: 2

TIME ALLOWED 2 ½ HOURS.

INSTRUCTIONS: ANSWER QUESTION 1 AND ANY OTHER 4 QUESTIONS

QUESTION 1

- a. Give reasons why methanol is an alcohol. **5 Marks**
- b. Define primary alcohols and polyhydric alcohols. **3 Marks**
- c. Discuss the properties of alcohol given below: **8 Marks**
 - i. Physical appearance
 - ii. Melting or boiling point
 - iii. Density
 - iv. Solubility
- d. Name and explain the process for preparation of primary, secondary and tertiary alcohols. **6 Marks**

QUESTION 2

- a. Organometallic reagents react with aldehydes and ketones to give alcohols. Give two examples of these organometallic reagents. **4 Marks**
- b. Give three enzymes that are involved in industrial production of ethanol. **4 Marks**
- c. List and arrange the classes of monohydric alcohol in order of increasing acidic, basic strength and order of reactivity involving R—OH cleavage. **4 Marks**

QUESTION 3

- a. List the types of ethers that are relevant to organic chemistry. **6 Marks**
- b. Ethers can be named using the common system or the IUPAC system. Fill in the missing names in the table below: **6 Marks**

c.

SN	STRUCTURE	IUPAC	Common Names
	NAMES CH_3OCH_3	Methoxy methane	
	$\text{CH}_3\text{CH}_2\text{OCH}_3$		
	$\text{CH}_3\text{CH}_2\text{OCH}_3$	Methoxy ethane	
	$\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_2\text{CH}_2\text{Cl}$		3-Chloropropyl ethyl ether
	$\text{CH}_3\text{CHCH}_2\text{CH}_3$ $\quad $ $\quad \text{OCH}_3$		2-Buthyl methyl ether

QUESTION 4

- Explain the concept of Metamerism in ethers. **6 Marks**
- List and discuss three methods of preparation of ethers. **6 Marks**

QUESTION 5

- List three each, physical and chemical properties of ethers. **6 Marks**
- Give two major group of compounds formed when ethers react at the ethereal oxygen.
6 Marks

QUESTION 6

- Define epoxide. **3 Marks**
- List the methods of formation of epoxides. **4 Marks**
- List all the possible reaction of epoxide. **5 Marks**