FBQ1: Agitation of a mixture of solution can be achieved using Answer: *Magnetic stirrer*
FBQ2: To avoid the errors in mass due to the use of balances that are not calibrated, one should weigh by a method called Answer: *Weighing by difference*
FBQ3: The mass of an empty beaker is 20.1672g; if the mass obtained after sodium trioxocarbonate (iv) was added to the beaker and weighed is 22.7587g, what is the mass of the sodium trioxocarbonate? Answer: *2.5915g*
FBQ4: is used in the laboratory to separate a mixture of two miscible liquids with different boiling points Answer: *Distillation*
FBQ5: Flammable solvents in a mixture should be boiled away in a Answer: *Fume cupboard*
FBQ6: is worn in the laboratory to avoid chemicals splashing into the eyes Answer: *Safety googles*
FBQ7: glassware is used to heat and evaporate liquids. Answer: *Evaporating dish*
FBQ8: Volumetric glassware is the most method of transferring and delivering liquids Answer: *Precise and accurate*
FBQ9: To prevent bumping of a hot liquid out of the container, is added. Answer: *boiling chip*
FBQ10: A metal sphere weighing 19.48g is added to 20ml of water in a graduated cylinder. If the density of the metal is 4.50g/ml, what will be the new level of water in the graduated cylinder? Answer: *24.33ml*
FBQ11: Reaction requiring low temperature of 00C can be carried out in the laboratory by employing Answer: *Ice water bath*
FBQ12: are used to crush solids into powders for experiments. Answer: *Mortar and pestle*
FBQ13: Tongs are used to hold apparatus when they are Answer: *Hot*
FBQ14: The apparatus below is calledÂ Answer: *Flat bottom flask*

FBQ15: The most basic technique for the purification of organic solids is
Answer: *Recrystallization*
FBQ16: are often used for heating in a chemical laboratory because they have low heat â€" resistance, they are less subject to thermal stress and are inert. Answer: *Borosilicate glasses*
FBQ17: Anhydrous compounds are kept dry in a Answer: *Dessicator*
FBQ18: Desiccant is a agent Answer: *Drying*
FBQ19: Bunsen burners are used for heating and exposing items to Answer: *Flame*
FBQ20: Filtration involves the separation ofsolid materials from a liquid. Answer: *Insoluble*
FBQ21: An impure substance melt atand over a wider range Answer: *A lower temperature *
FBQ22: The colour of bromine solution turns alkene compound to Answer: *Colourless*
FBQ23: The loss of water from a molecule is called Answer: *Dehydration*
FBQ24: A is used to contain/put a small quantity of substance to be heated to a very high temperature? Answer: *Crucible*
FBQ25: Cycloalkane are called saturated hydrocarbon because of Answer: *single bond*
FBQ26: Domestic scale is a type of weighing scale. Answer: *Digital*
FBQ27: The balance which display its mass reading in grams to 2 decimal place is called Answer: *Top loading balance*
FBQ28: type of distillation is based on large number of theoretical vaporization-condensation cycle. Answer: *Fractional distillation*
FBQ29: Separation techniques involving distillation obey type of law.

Answer: *Raoult's law*
FBQ30: Â-Â-Â-Â-Â-Â-Â-Â-Â-Â-Â-Â is based on the principle of the equilibrium distribution between two immiscible phases. Answer: *Extraction*
FBQ31: Â The object shown above is called Answer: *Wash bottle*
FBQ32: The mass of a substance divided by the volume of the substance is known as
Answer: *Density*
FBQ33: Vinegar contains a carboxylic acid called Answer: *Acetic acid*
FBQ34: The temperature of an object/substance can be determined using a
Answer: *Thermometer*
FBQ35: A mixture of sand and water can be separated from each other by Filtration, decantation and Answer: *Evaporation*
Multiple Choice Questions (MCQs): MCQ1: Theis used in vacuum or suction filtration in order toseparate solids from liquids. Answer: Evaporating dish
MCQ2: Which of the following is used to hold solids when being weighed orTransported? Answer: Conical flask
MCQ3: During chemical reactions in the laboratory, reagents are stirred so as to
Answer: Mix the reagents or initiate the reaction
MCQ4: is a more accurate weighing balance amongst the options listed below. Answer: Analytical balance
MCQ5: All of the options below are part of a sample distillation apparatus exept Answer: Heating mantle
MCQ6: is not a laboratory heating device. Answer: Hot plate
MCQ7: is used for heating solutions up to 1000C and not more.

Answer: Water bath MCQ8: Which weighing balance will you use if a procedure tells you to weigh the sample accurately? Answer: Analytical balance MCQ9: Which of the following separation technique does not give a pure product? Answer: Simple distillation MCQ10: The glassware above is used for ______. Answer: temporary storage MCQ11: Which of the following will introduce error when weighing an object or substance? Answer: Placing the weighing balance in a location with low levels of vibration and air current MCQ12: Liquids are separated from solids with the following except _____. Answer: Decanting MCQ13: What would you do if you were to heat a sample in a dish? Answer: Weigh it immediately MCQ14: . Most fire in the laboratory can be prevented by the use of _____. Answer: Extinguisher MCQ15: The following are the first things to be done when determining the density of an unknown metal cube except _____. Answer: Weigh the metal cube MCQ16: Using an unclean volumetric glassware during experiment will ______. Answer: Slow the rate of the reaction MCQ17: Amongst the options listed below ______ is a better choice for the heating of flammable substances. Answer: Evaporating dish MCQ18: What is the first thing to do when a person inhales vapour of an irritating or toxic substance? Answer: The person should be taken away immediately to fresh air first MCQ19: Vaporization and condensation are involved in _____ processes. Answer: Simple distillation MCQ20: Â This apparatus is used for _____. Answer: For transfer of liquids accurate method of transferring and delivering liquids.Â

Answer: Medicine droppers
MCQ22: is not a separation technique frequently employed in the laboratory to isolate one or more components from a mixture. Answer: Recrystallisation
MCQ23: Which of these statement is true? Answer: Fractional distillation involves one cycle of vaporisation - condensation
MCQ24: A graduated cylinder is filled to the 40.00 ml mark with mineral oil. The masses of the cylinder before and after the addition of mineral oil are 124.966 g and 159.446 g. Determine the density of the mineral oil. Answer: 34.48 g/ml
MCQ25: amongst the options is not used in gravity filtration? Answer: Beaker
MCQ26: is not a volumetric glassware. Answer: Round bottom flask
MCQ27: A chemist would determine several physical and chemical properties of a compound because Answer: he wants to know more about the compound
MCQ28: Which of the following will you do when a chemical accidently get into your eye? Answer: Clean your eye with a clean cloth
MCQ29: Compound A was found to have a higher molecular weight than Compound B. Which of them will have a higher boiling point? Answer: Compound B
MCQ30: What does a foreign particle do to a crystal lattice? Answer: It interrupts its uniform structure
MCQ31: Which of these is/are more accurate and precise in taking weight measurements?Â Answer: Iron Scale
MCQ32: The process that boil reactant while continually cooling the vapour back to the flask as liquid is called Answer: Condensation
MCQ33: The type of distillation that shows more precision and efficiency is Answer: Simple distillation
MCQ34: The commonest type of stirrer used for a reaction mixtures that is very viscous is Answer: Magnetic stirrer

MCQ35: Which weighing balance will you use if a procedure tells you to weigh about 3g of ammonium sulphate?

Answer: Analytical balance